On the Structure of *Trans* Ethyl Methyl Ether and the W Form of Diethyl Ether

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We have measured the microwave spectra of ethyl methyl ether and its isotopically substituted species. Since the measurements have not been finished for 18 O and one of 13 C species, the r_s structure can not be obtained at present.

We found many weak spectra around the expected regions for the spectra of the ¹⁸O and one of three ¹³C species. Some of these spectra may be assigned to those of the vibrationally excited states of the molecule. Since these spectra are confusing, it is preferable to assign them before the assignments of the spectra are undertaken for the ¹⁸O and the ¹³C species. However, we have tried to determine a tentative structure of *trans* ethyl methyl ether which reproduces the observed rotational constants for the parent and nine isotopic species.

In Table 1 are shown the observed rotational constants and the differences between the observed and the calculated rotational constants which were calculated

using the obtained structural parameters given in Table 2. Work is in progress with respect to both the *trans* and the *gauche* isomers on the electric dipole moment, the internal rotation of the CH₃CH₂ and OCH₃ groups and the skeletal torsion and spectra due to the vibrationally excited states.

We have also tried to measure the microwave spectra of diethyl ether based on the trial calculation for the rotational constants of four possible molecular forms of diethyl ether using the structural parameters given in Table 2. We could find a set of spectra assigned to the W form (or the trans-trans form) of the molecule. In order to reproduce the observed rotational constants, we have adjusted only the structural parameters of the skeleton. The best agreement is obtained when we reduce the \angle COC angle by about 1° from the values for trans ethyl methyl ether. The results are given in Table 1.

Table 1. The observed rotational constants (MHz) and the differences between the observed and predicted rotational constants

Species		$A(\varDelta A)^{\mathrm{a})}$	$B(\Delta B)^{\mathrm{a}}$	$C(\Delta C_i)^{\mathrm{a}}$	
trans ethy	l methyl ether				
$CH_3CH_2OCH_3$		27992.03 (13.96)	4159.61 (4.61)	3891.11 (0.06)	
CH ₃ CH ₂ OCH ₂ D	$\mathrm{sym^{b}}$	27218.96 (5.98)	3956.97 (4.21)	3699.99 (-0.74)	
	asym ^{b)}	26194.77 (0.52)	4009.66 (6.49)	3765.02 (2.23)	
CH ₃ CHDOCH ₃		25040.20 (10.71)	4131.86 (4.86)	3843.20 (0.18)	
CH ₃ CD ₂ OCH ₃		22669.17 (10.70)	4095.93 (5.23)	3804.65 (0.05)	
CH ₂ DCH ₂ OCH ₃	$\operatorname{sym}^{\mathrm{b})}$	27505.31 (7.36)	3937.57 (-0.39)	3683.49 (0.02)	
	asym ^{b)}	26057.29 (18.57)	4008.84 (4.44)	3760.41 (0.20)	
$CH_3CH_2OCD_3$		23838.31 (0.85)	3704.12 (7.48)	3491.59 (3.02)	
¹³ CH ₃ CH ₂ OCH ₃ ^{c)}		27951.36 (17.20)	4041.94 (6.71)	3787.16 (2.23)	
CH ₃ CH ₂ O ¹³ CH ₃ c)		27978.56 (16.60)	4051.52 (6.66)	3796.06 (2.16)	
the W fo	rm of diethyl e	ther			
$(CH_3CH_2)_2O$		17955.83 (6.87) ^{d)}	$2244.15 (2.71)^{d}$	2101.73 (0.12) d	
-		$(-114.65)^{\circ}$	$(-0.27)^{\mathrm{e}}$	$(-4.15)^{e}$	

- a) $\Delta R = R_{\text{calctd}} R_{\text{obsd}}, R = A, B, C$
- b) "sym" and "asym" refer to the symmetric and asymmetric forms, respectively with respect to the molecular plane.
- c) Since the Q branches of the transitions have only been measured at present for two ¹³C species, the rotational constants are determined on the assumption that the difference of $I_a + I_b I_c$ between the parent and the isotopic species is the same as that found in CH_3CH_2Br and $(CH_3)_2O$, respectively.
- d) The difference between the observed and predicted rotational constants calculated from the adjusted structural parameters.
- e) The difference between the observed and predicted rotational constants calculated from the structural parameters obtained for trans ethyl methyl ether.

Table 2. Structural parameters obtained for trans ethyl methyl ether

$\mathrm{CH_3}$	O	CH,	$_{3}$ C	CH	\mathbf{H}_2	Skele	ton
$r(CH_s)$	1.091Å	$r(\mathrm{CH_s})$	1.083Å	r(CH)	1.097Å	r(CC)	1.540Å
$r(CH_a)$	1.097Å	$r(CH_a)$	1.087Å			$r(\mathrm{CH_3O})$	1.412Å
$\angle H_aCH_a$	109°24′	$\angle H_aCH_a$	108°48′	\angle HCH	107°30′	$r(\mathrm{CH_2O})$	1.412Å
$\angle H_aCH_s$	110°10′	$\angle H_aCH_s$	109°45′	\angle HCC	110°51′	$\angle COC$	111°23′
$\angle H_{\mathfrak{s}}CO$	10 7 °13′	$\angle \mathrm{H_sCC}$	109°30′	\angle HCO	109°50′	∠occ	107°57′
$\angle H_aCO$	110°5′	$\angle H_aCC$	109°30′				

locked angles for: CH₃O, 1°55'; CH₂O, 1°15'